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What Is An Alloy? 4. Identify The Following Compounds As Metallic, Ionic Or Covalent: A. NaCl B. Cl<sub>2</sub> C. Au D. [BrO<sub>3</sub>]<sup>-1</sup> E. Mg<sub>3</sub>N<sub>2</sub> F ... 8th, 2021

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## **Nomenclature #1: Binary Ionic Compounds**

Nomenclature #2: Polyatomic Ionic Compounds 1.

Name The Following Compounds (include Roman Numerals When Necessary):  
Na<sub>2</sub>SO<sub>4</sub> Sodium Sulfate  
AlPO<sub>4</sub> Aluminum Phosphate  
Al(ClO<sub>4</sub>)<sub>3</sub> Aluminum Perchlorate  
AsPO<sub>3</sub> Arsenic (III) Phosphite  
Ni(OH)<sub>2</sub> Nickel (II) Hydroxide  
AgBrO<sub>3</sub> Silver Bromate  
Pb(IO<sub>3</sub>)<sub>2</sub> Lead (II) Iodate  
K<sub>3</sub>P Potassium Phosphide  
HgCN Mercury (I) Cyanide  
Mg(IO<sub>4</sub>)<sub>3</sub> 3th, 2021

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